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Leachate Monitoring in Naturally Saline Groundwater Chesapeake Landfill Chesapeake Virginia

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LEACHATE MONITORING IN NATURALLY SALINE GROUNDWATER, CHESAPEAKE LANDFILL,

CHESAPEAKE, VIRGINIA

by

T. Britt McMillan R.A. May 1981, Old Dominion University

^A Thesis Submitted to the Faculty of Old Dominion University in Partial Fulfillment of the Requirements for the Degree of

MASTER OF SCIENCE

GEOLOGY

OLD DOMINION UNIVERSITY December, 1985

Approved by:

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Dennis A. Darby

G. Richard Whittecar

ABSTRACT

LEACHATE MONITORING IN NATURALLY SALINE GROUNDWATER, CHESAPEAKE LANDFILL, CHESAPEAKE, VIRGINIA

T. Britt McMillan Old Dominion University, ¹⁹⁸⁵ Director: Dr. J. H. Rule

Groundwater chemistry around the Chesapeake municipal landfill was monitored over ^a one year period. Ten sample sites as well as two surface water sites were used to monitor water quality. Two wells, one at ³ ^m and the other at ¹⁰ ^m were located at each site. Surface water samples were taken from the Elizabeth River, north of the landfill, and ^a tidal channel, west of the landfill. Seven groundwater sites were downgradient of the landfill and three sites were upgradient (control sites).

The landfill overlies ^a tidal marsh, approximately 100 ^m south of the intracoastal waterway (Elizabeth River). Dredge spoils overlying a marsh clay-muck separate the landfill from the waterway to the north. To the east and south is ^a sandy loam soil and to the west is a tidal marsh. The underlying aquifer is fairly homogeneous vertically and horizontally, consisting of medium to fine, moderately sorted sand which is strongly fine-skewed leptokurtic.

Groundwater and surface water samples were monitored for pH , Eh, temperature, conductivity, salinity, hardness, NO_3 , $NO₂$, TKN, TPO₄, OPO₄, SO₄, C1, Na, K, Ca, Mg, Fe, Mn, and Zn. ANOUA and factor analysis aided in identifying sources of variance in the parameters measured. Conductivity, salinity, hardness, sulfate, sodium, chloride, and magnesium, though present in high concentrations in the leachate, were most indicative of the surface water. Potassium, total and orthophosphate, and TKN best characterized the leachate.

Tidal fluctuation had no observable impact on the groundwater chemistry, though there did appear to be some seasonal influence on the leachate concentration.

ACKNOWLEDGEMENTS

I would like to express my appreciation to all the people who gave guidance and assistance during this investigation. ^I would like to thank Dr. Joseph H. Rule, my thesis director, without his invaluable supervision and direction this work would never have been completed. To Dr. G. Richard Whittecar, his assistance, both in the field and in the office is greatly appreciated. His insight and suggestions on many aspects of this work was especially helpful. Dr. Dennis A. Darby's careful review, and his valuable criticisms and suggestions regarding this work has been most appreciated.

I would like to thank the City of Chesapeake for funding the installation of monitoring wells, and thanks to Mr. Jim Garrett for assistance in the drilling operation.

^I would like to express my gratitude to the faculty, staff, and graduate students of the Geological Science Department, especially those who, at one time or another, aided me with my field or laboratory work. Special thanks to Mr. Charlie Fox, Ms. Jean Ashmore, and Ms. Linda Ruf for their assistance over the years.

^I am greatly indebted to my parents for their unwavering support they have given me. ^I can never express my full gratitude for all they have done.

To my wife, Velja McMillan, who not only has tolerated the long hours ^I have devoted to this research, but has actively participated in all of its aspects. She has

ii

enthusiastically worked both in the field and laboratory with me, and is the typist of this document. Her willingness to take on many of the responsibilities of running ^a household enabled me to have more time to work on my research is a sign of her love and committment to me.

TABLE OF CONTENTS

TABLE OF CONTENTS (CONT'D)

PAGE

Appendixes

LIST OF TABLES

 $\ddot{\ddot{\mathbf{v}}}$

LIST OF FIGURES

LIST OF FIGURES (CONT'D)

INTRODUCTION

Sanitary landfills and open dumps have been and still are the most widely used methods for disposal of municipal solid waste (MSW). The sanitary landfill, introduced in the 1930's, is considered the safest, most efficient method for land-based disposal of solid waste. However, many studies in recent years have demonstrated the landfill's potential for degradation of groundwater quality around the landfill (Qasim and Burchinal, 1970; Fungaroli, 1971; Chain and DeWalle, 1976; Johansen and Cocozza, 1977; Landreth, 1978; Gibb et al., 1981; Lu et al., 1981). The majority of these studies involved landfills with their bases in the unsaturated zone, which tends to restrict movement of the leachate.

Composition and volume of leachate generated by landfills is highly unpredictable due to variations in landfill design, operation, and stabilization. The following factors are most important in determining the composition and volume of leachate generated:

- 1. landfill age
2. waste compos
- 2. waste composition
3. landfill design am
- 3. landfill design and operation
4. local climate
- 4. local climate
5. local hydrolo
- 5. local hydrologic conditions
6. characteristics of the under
- characteristics of the underlying soil or sediment

Of these factors, landfill age has the greatest influence on leacheate composition (gasim and Burchinal, 1970; Chain and

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DeWalle, 1976; Johnansen and Carlson, 1976; Lu et al., 1981).

The major constitutents of most MSW disposal facilities are paper and other wood products, vegetable matter, animal wastes, metal, glass, and ash. The principle pollutants from these wastes are soluble organic and nitrogenous compounds. These contaminants are typically measured as Biological Oxygen Demand (BOD), or Total Organic Carbon (TOC) plus Chemical Oxygen Demand (COD), and Total Kjeldahl Nitrogen (TKN). In addition to the organic compounds, a host of inorganic ions are commonly found in leachate. Ions of relatively low toxicity include: Na, K, Ca, Mg, Mn, Zn, Fe, NH₄, Cl, SO₄, PO₄, and HCO₃. Pb, Ni, Cu, Cd, Ba, Hg, Cr, B, CN, F, $NO₃$, As, and Se are ions of relatively high toxicity (many of which are site specific). ^A complete listing of parameters used as leachate indicators is in Table 1.

Effects of landfill age for several of these parameters are well summarized by Chain and DeWalle, (1976, 1977). Decreases in ratios of COD/TOC, BOD/TOC, and $SO_4/C1$ with age reflect changes in organic matter composition. These decreasing ratio trends and increases in pH and Eh result from rapid biodegration of free volatile fatty acids, leaving relatively stable, high molecular weight carbohydrate complexes and inorganic ions.

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TABLE 1. Leachate Indicators (Penn and Cocozza, 1977)

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consequence, landfills in these regions are often located in or adjacent to coastal marshlands. These site locations present a monitoring problem as well as causing degradation of groundwater and estuarine waters (MacGregor et al., 1980; Lee et al., 1982).

Complications arise when pH and chloride are used as leachate indicators in ^a tidal marsh situation. Change in pH or increase in Cl from ambient groundwater concentrations due to leachate would be indistinguishable from intrusion of saline water from the ocean or tidal channels. Little, if any, research has been published on the movement and effects of leachate in coastal marshlands.

FEDERAL REGULATIONS

Until recently, wetlands (both fresh and saline) were either filled with dredge spoil to make the land suitable for development or used as an economically attractive site for disposal of both solid and liquid wastes. Much of the nation's wetlands has been destroyed or adversly impacted by such use. Land use within or adjacent to wetlands has been only loosely regulated at the Federal, State, and local levels. However, in the past ²⁰ years, all levels of government have begun to show increasing concern over the nation's wetlands. One of the most significant steps at the federal level to regulate land use around wetlands was passage of the Resource Conservation and Recovery Act (RCRA) in 1976. The RCRA provides for direct, centralized regulation of all solid waste disposal in the United States

 5

under joint Federal and State control. This act is to be administered by the Environmental Protection Agency (EPA) pending completion of regulations and guidelines. Another major step toward regulation of landuse around wetlands was the 1977 revision of Section 404 of the Clean Water Act. In this revision, jurisdiction over permits for dredging and filling in wetlands was granted to the U.S. Army Corps of Engineers. Implementation of the 404 permit program is presently pending completion of the EPA's wetland impact assessment. Problems yet to be resolved are prediction of landuse impacts on wetlands, individually as well as in conjunction with other activities; assessment of impacts on an area-wide versus site-specific basis; and assessment of impacts from exempt or unregulated activities around wetlands (Thibodeau, 1981; U.S. EPA, 1979; MacGregor et al., 1980; Nelson, 1983).

PURPOSE

The objective of this study is to establish the suitability of pH and chloride as leachate indicators in groundwaters with naturally high salinities. If these parameters prove to be unsuitable, applicability of other selected parameters will be evaluated. Those which best fit the criteria stated in the introduction for potential leachate indicators will be considered the most viable alternatives to pH and chloride for routine monitoring purposes.

In order to meet this objective, factors in addition to groundwater geochemistry surrounding the Chesapeake Landfill are considered. An approximation of the transmissivity and direction of groundwater flow as well as grain size and thickness of the aquifer influenced by the landfill were evaluated. Tidal fluctuation and seasonal change were examined, as well, in order to evaluate their influence on those parameters studied.

STUDY SITE

The Chesapeake Landfill is ^a municipal solid waste disposal facility located in the southeastern coastal plain of Virginia and has been in operation since the mid 1960's. The site overlies ^a tidal marsh, approximately ¹⁰⁰ meters south of the Intracoastal Waterway. Dredge spoils separate the landfill from the waterway to the north. To the east and south is a sandy loam soil and to the west is ^a tidal marsh (Figure $1 \& 2$). The dredge spoil is a medium-sorted, fine sand directly overlying the marsh clay-muck. An abundance of shell fragments and a high concentration of iron oxide-coated sand are found adjacent to the landfill. The dredge material is two meters thick wear the landfill tapers to one meter toward the waterway. ^A berm consisting of boulders and dredge spoil lines ^a portion of the river bank.

The water table aquifer ranges from less than a meter to three meters below the surface and extends seven to ten meters in depth where it contacts the Chowan River Formation.

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Figure 1. Map of Study Area in relation to the regional geography.

Figure 2. Map of landfill and adjacent areas showing the distribution of the soil types (from Henry et al 1958).

Between the landfill and waterway this aquifer is semiconfined, separated from the dredge spoil by marsh clay. General direction of groundwater flow is north, toward the waterway. The water table aquifer consists of fine to medium sand, generally increasing in size with depth (Appendix A). Shells are abundant in this unit, with greater concentrations toward the base. Transmissivity values for this aquifer range from 1,400 gpd/ft to 2,600 gpd/ft (Siudyla et al., 1981).

The landfill base is located several feet beneath the water table. The original mode of operation was to trench and dewater while refuse was deposited and compacted. Initial dumping was in the eastern portion of the landfill, with progressive filling toward the west. Presently, refuse is being placed over the older portion of the landfill. Two wells, placed at ⁷ 1/2 meters depth, are currently being monitored by the city at irregular intervals for leachate. The parameters measured for these wells are pH and Cl. Several studies have suggested that salt water intrusion from the waterway may tend to mask high chloride levels due to leachate in the groundwater (Virginia State Water Control Board, unpublished data; Rule, 1979; Rule, unpublished data; McMillan, 1981).

PREVIOUS STUDIES

Studies of the Chesapeake Landfill conducted by Rule (1979) involved eight sample sites (Figure 3). The samples were taken by peristaltic pump and tygon tubing. Levels of

Figure 3. Sample locations used ^b^y Rule (1979).

pH and Eh were determined in the field and ihe samples for metal analysis were field-filtered through ^a 0.45 micron membrane, then acidified with 1:1 HNO3. Coliform samples were cooled on ice and planted within six hours after sampling. The samples for metals analysis were digested in the lab using distilled HNO3 and reagent grade HCL, in accordance with EPA methods (U.S. Environmental Protection Agency, 1974). Thirteen parameters were measured: pH, Eh, total and dissolved solids, total and fecal coliforms, Cl, Cd, Cr, Cu, Ni, Pb, and Zn. For the metals, both total and dissolved concentrations were determined. The results of the analyses indicated several monitoring problems. The monitoring wells were cased in galvanized metal, which could potentially produce anomalously high metal concentrations (note concentration of Zn, Appendix 8, sites ¹ and 3). The study also showed high Cl levels near the canal indicating possible saltwater intrusion (Appendix 8, site 3). If the salinity of the groundwater was greater than the leachate, a density-separated flow would result in which leachate would flow above the monitoring well points. In addition, naturally high Cl levels in ihe groundwater would tend to mask Cl levels in ihe leachate.

The vertical positions of the wells (sites ¹ and 3) within the aquifer may also present problems in leachate monitoring. The well points were placed at a depth of ⁸ meters. ^A well log is not available, so the positions of

the well points relative to the base of the aquifer are unknown.

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Preliminary research conducted by McMillan (1981) involved installation of eight pressure-vaccuum lysimeters in a transect between the landfill and waterway. Two lysimeters were installed at each well site, at depths of one and three meters (Figure 4). No control wells were used in this study. The wells were all hand-augered. Due to hydraulic pressure and incompetency of the sand it was impossible to auger deeper than three meters. The three meter well at site ^D did not penetrate through the clay layer, so no sample could be obtained from this lysimeter. There were five sampling periods from March through July 1981. Eh, pH, conductivity and salinity were determined in the field. Samples taken for metal analysis were fieldpreserved with $1:1$ HNO₃, the other samples received either no preservative or H_2SO_4 , and were cooled to four degrees centigrade. All analyses were conducted in accordance with EPA Methods (U.S. Environmental Protection Agency, 1974). The parameters measured in this study were pH, Eh, conductivity, salinity, TKN, $NO₃$, $NO₂$, Cl, total $PO₄$, and metals (Na, K, Ca, Mg, Fe, Mn, Cu, and Pb).

The following parameters showed distinctive trends (Appendix 8):

1. Conductivity, salinity, Na, Cl, and Mg concen- trations for the upper wells showed ^a general decrease away from the waterway, increasing again toward the landfill. The concentrations increased from the Elizabeth River toward the landfill for the deeper wells.

- 2. Concentrations of TKN, K, Fe, and Ca increased toward the landfill for both upper and deeper wells.
- 3. NO₂ increased by several orders of magnitude toward
the landfill for the one meter wells. The only significant concentration for the three meter wells
is at site C (no sample was obtained at three meters for site D).
- 4. PO₄ was the only parameter to decrease in concen-
tration toward the landfill for the one meter
wells. Concentrations for the three meter wells were insignificant when compared with the one meter
wells.

Results from the preliminary study by McMillan, (1981), tentatively indicated that influence from the canal resulted in high salinity, conductivity, Na, Cl, and Mg in the dredge spoil. In addition to the parameters above, leachate from the landfill appeared to contribute high TKN, K, Fe, NO₃ and Ca concentrations for both the dredge spoil and the water table aquifer. ^A portion of the Ca concentration for the upper wells may be attributed to the shell fragments present in the sediment at sites ^C and D. However, no shells were found in the water table aquifer even though high Ca concentrations were present.

METHODS AND PROCEDURES

FIELD METHODS AND LABORATORY PROCEDURES

The monitoring wells for the present study consist of 1-1/4 inch PVC pipe with a three foot fine screened well point. These wells were installed by ^a wash boring rig, backfilled with sand from the aquifer and sealed at the top with bentonite. Due to problems with collapsing sand, the ten meter wells were jetted in and as ^a result are not capped at the bottom of the screen. Logs were taken for each well site (Appendix A). Elevations of the top of all wells were measured by transit and stadia rod.

Each site has ^a well positioned at the upper (three meters depth) and lower (ten meters depth) boundary of the aquifer. Upper wells are designated by a subscripted ¹ and lower wells by ^a subscripted 2. Two parallel transects are located to the north of the landfill (Figure 5). Three well sites per transect are spaced at approximately twenty five meter intervals, from river to landfill. An additional monitoring site is installed to the west of the older portion of the landfill (site G), as well as three control sites (H, I, J) to the east and south. Additional water samples were taken from the tidal creek adjacent to site ^G and from the river, near site A.

Figure 5. Locations of well sites and surface water sites.

Changes in hydraulic head, conductivity, salinity, and temperature for sites A, C, and ^I were measured hourly over ^a ³⁰ hour period in October 1982. Water levels in all wells were measured in October 1982 and August 1983.

Sample collection and preservation were conducted in accordance with EPA recommended procedures (Penn et al., 1977; U.S. EPA, 1979; Gibb et at., 1981). Samples were withdrawn using ^a peristaltic pump, after drawing off at least 15 liters from each well to insure a fresh sample. Due to very slow recharge for wells E_2 and I_2 , there was no initial flushing of these wells. Samples were stored in one-liter linear polyethylene (LPE) bottles and immediately placed on ice. Separate samples in ⁶⁰ ml bottles were taken for pH and Eh and measured on site. Conductivity, salinity, and temperature were taken by lowering a conductivity cell and temperature thermistor probe in each well after samples were obtained. Within ²⁴ hours after sampling, the samples were centrifuged at 6,000 rpm for five minutes to remove suspended particles. Samples for metal analysis were then stored in ¹⁵⁰ ml LPE bottles and preserved with redistilled reagent grade HN03 at ^a pH of less than 2. Samples to be analyzed for phosphates, nitrate, and Total Kjeldahl Nitrogen (TKN) were stored in 250 ml bottles and preserved with reagent grade H_2SO_4 at a pH of less than 2. Samples preserved with either H_2SO_4 or no preservative were stored at a temperature of 4 degrees centigrade.

The parameters measured were pH, Eh, conductivity, salinity, hardness, temperature, TPO_4 , OPO_4 , NO_3 , NO_2 , TKN, SO4, Cl, and metals (Ca, Na, Mg, Mn, K, Zn, and Fe). Samples were taken on a monthly basis from August 1982 through October 1983, for a total of twelve sampling periods. ^A Ag Ag(CI combination electrode was used to measure pH; a platinum redox electrode for Eh; and conductivity, salinity, and temperature was measured by a YSI model 33 SCT meter. Both total and orthophosphate were determined by the ascorbic acid method, with a persulfate digestion prior to addition of the coloring reagent for total phosphate. Nitrate was measured by the brucine method, nitrite by the sulfanilamide method, TKN by digestion and ammonia probe, sulfate through the barium chloride turbidometric method, and chloride by either solid state electrode or argentometric method. Metals were determined with a Perkin Elmer 603 atomic absorption spectrophotometer.

The most ubiquitous interferences were highly colored samples from sites C_1 , F_1 , G_1 , and G_2 and colloidal suspension (primarily from sites E_2 and I_2). Parameters which relied on spectrophotometric methods (nitrate, nitrite, total phosphate, and orthophosphate) or turbidimetric methods (sulfate) in determining their concentrations were affected most. To correct for these interferences for nitrate, duplicate samples were digested without the coloring reagent, and used as blanks. For nitrite, total phosphate, and orthophosphate, initial absorbances were

read before addition of the coloring reagents and used as blanks. Interferences from color and colloids were corrected for sulfate by initial absorbance readings, after addition to the conditioning reagent and before addition of the barium chloride. The standard addition method was used for several samples during most analyses to verify that any interferences present were not significant. In addition to standard additions, EPA quality control samples were used for most parameters for at least one sampling period.

Samples were stored and preserved in accordance with EPA recommended procedures (U.S. Environmental Protection Agency, 1979). All analyses were conducted in accordance with Standard Methods (APHA-AWWA-APCF, 1975) and within the alloted sample holding time as specificed by EPA methods.

STATISTICAL METHODS

Statistical evaluation of the data was divided into three parts. First, variance within the data was discussed utilizing descriptive statistics and one-way analysis of variance (ANOVA). Second, multiple regression was used to examine any possible relationship between tidal fluctuation and variance within a parameter. If tidal fluctuation was found to significantly influence a parameter, the regression equation was used to correct for this influence. Third, factor analysis was used to summarize the interrelationships among the variables, condensing the variance within the original data into a few variables (factors) as an aid in conceptualization. The statistical package SAS (Statistical

Analysis System) compiled by SAS Institute Inc., was used to obtain solutions to the ANOVA, multiple regression, and factor models.

Before results from the ANOVA could be interpreted, potential failure of two basic assumptions had to be considered: within-cell observations are normally distributed about the mean; and variance between cell means is homogeneous. The Barlett-Box F-statistic was used to test homogeneity of variance. Often, when non-normal distributions occur, heterogeneity of variance between means (heteroscedastcity) also occur (Cochran, 1947). This non-normal distribution and heterogeneity of variance errors are usually ^a direct function of the cell's mean value $(S^2=m+S^2_m)$. A lognormal transformation may be used to correct this failure. Barlett (1947) considered this to be the appropriate transformation for non-normal sample variances. If the lognormal transform significantly improves the distribution of error terms, then the transformed data would be used in all subsequent analyses. This transformation has been widely used in geochemistry to correct for pseudo lognormal distributions, however, application of this method is still in dispute (Link and Koch, 1975, Chapman, 1976; 1977; Miesh, 1977).

Because the data for the ANOVA model was not a balanced design, a general linear model was used. This model is considered a good alternative to the more traditional method of mean square ratios where the cell block design is

unbalanced (Wesolowsky, 1976; Snedecor and Cochran, 1980). Tukey's range test was used to aid in identifying anomalously high or low cells (sample sites or dates) for parameters where the null hypothesis was rejected.

The independent variables used for the multiple regression analysis was tidal fluctuation, seasonal variation, and horizontal and vertical distance. Tidal levels were taken from tide tables for Sewells Point, Norfolk, and corrected for the Great Bridge locks. Tidal fluctuation was recorded as a fraction ranging from 0.0 to 1.0, with 0.0 representing low tide; 1.0 equal to high tide; and 0.5 as slack tide. Seasonal variation (summer, winter, spring, and fall) was represented by three dummy variables. Distances were also recorded as dummy variables, iwo for depth (upper wells, lower wells, and surface water) and four for horizontal distance (surface water, adjacent to landfill, adjacent to surface water, between landfill and surface water, and control wells).

The SAS procedure REG was used for ^a least-squares fit to the regression models. All independent variables were used as regressors in the first model. Subsequent models used separate dummy variable groups (seasonal variation, vertical distance, and horizontal distance) as the regressor variables.

The approach to factor analysis of ihe groundwater data was to use the most simplistic model (principal components) and derive principal factors (or axes) and scores for the

total data set. The loadings for each factor were then compared to the corresponding score groupings. If of the scores into groups was geologically
ble in relation to their loadings, a higher interpretable in relation to their loadings, ^a higher level of factor analysis was employed. The method chosen was a principle axes solution with varimax rotation (vector analysis). Results from this analysis were then compared with the principle component solution to see if it increased resolution without changing the basic distribution of the factor scores.

Standarized data was used to calculate factor scores, therefore the sum of all observations for each variable has a mean of zero and unit variance. The scores were grouped according to their original sample sites and plotted as bargraphs with 95% confidence bands for each site.

Pairwise deletion of observations was used to produce the correlation matrix. For calculating factor scores, only observations with missing variables contributing to less than 10% of the vector's magnitude were used. Missing values for included observations were replaced by sample means. Since these values have little influence on the resulting score, this method was considered best for retaining ^a maximum amount of information with mininal sacrifice to error.

RESULTS AND DISCUSSION

STRATIGRAPHY AND HYDROGEOLOGY

The landfill overlies tidal marsh deposits approximately 100m south of the intracoastal waterway (southern branch of the Elizabeth River). Dredge spoils overlying ^a marsh clay-much separate the landfill from the waterway to the north. To the east and south is a sandy loam soil and to the west is ^a tidal marsh. The stratigraphy of the water table aquifer is known from wash boring logs taken when the monitoring wells were installed and from particle size analyses of sediments taken from two sites north of the landfill. These data form the basis for a generalized fence diagram (Figure 6) and a more detailed cross-section (Figure 7).

The underlying aquifer is fairly homogeneous both vertically and horizontally. It consists of medium to fine, moderately sorted sand and is strongly fine-skewed and leptokurtic. Parallel to the northern edge of the landfill and extending no more than 50m north of the landfill is ^a ³⁰ to ⁶⁰ cm thick silt-clay lens at ^a depth of seven meters. This lens was recorded in well logs at sites C, ^F and G, and borehole ¹ (Figures ⁶ and 7). Shell fragments are noticeably more abundant with depth. From the well logs, sediments around control wells H, I and J

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Generalized fence diagram connecting all Figure 6. boreholes around the Chesapeake Landfill.

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appear to be homogeneous both vertically and horizontally, consisting of medium-fine sand.

Water levels in the wells seem to reflect the degree of variability in sediment textures. For example, most wells refilled with water almost instantly when purged during water sampling. Also, differences in water levels between upper and lower wells (verical hydraulic gradient) was constant and relatively small (0.01) between most well sites. Both of these observations indicaate that the aquifer is reasonably homogeneous with ^a high permeability. Not all well sites, however, are so uniform. At wells I2 and E2, refill rates are much slower than at other sites and there is a greater decrease in hydraulic head from upper (3 meters) to lower well (10 meters). Well F_1 also refilled noticably slower than most wells, though not to the same extent as E_2 and I_2 . The slow refill rates at wells E₂ and I₂ and the high vertical gradients at these sites are most likely due to a decrease in grain size with depth. This would indicate heterogeneity in the sediment texture is greater than indicated by the wash boring logs.

Due to the general lack of regional topographic relief and the gentle, broad slopes of coastal terraces in the area, it was assumed prior to this study that groundwater flow is generally north and the hydraulic gradient is low. In order to test this assumption, water levels were measured for all wells in October, ¹⁹⁸² at the same point in the tidal cycle (Appendix A). Wells along the N-S transect A-I

were monitored semi-hourly over a 30 hour period in October, 1982. Use of piezometers rather than wells screened through the entire aquifer presented a problem in evaluating the overall horizontal gradient in that the measured water levels were influenced by vertical hydraulic gradients in addition to horizontal gradients. The horizontal gradient for the upper wells is generally north (NW to NE), toward the river, averaging around 0.005. There is almost no horizontal gradient for the lower (10 meter) wells (0.001) . The vertical gradient for all sites (except J) decreased with depth. ^A three dimensional hydrologic cross-section shows the head distributuion in the aquifer north landfill (Figure 8). Effects of groundwater h depth. A three dimensional hydrologic cross-section
ws the head distributuion in the aquifer north of
landfill (Figure 8). Effects of groundwater mounding in
eastern half of the landfill and the apparent decrease
grain s the eastern half of the landfill and the apparent decrease in grain size toward the west is primarily responsible for the cone, or plume shaped distribution centered around transect A-C. The higher mounding along transect A-C is due to the topographically higher (5 to ⁷ meters) elevation of the eastern portion of the landfill over the western half. This area is also currently active (unvegetated), with sandy dredge spoil used for cover.

Water levels measured semi-hourly over ^a thirty hour period along the north-south transect A-I show a decrease in the hydraulic gradient toward the river (Figure 9). The lateral gradient is much higher north of the landfill than elsewhere due to effects of groundwater mounding in the landfill. Tidal fluctuation in the adjacent river

Figure 8. Three dimensional hydrogeologic cross section of the head distribution north of the Chesapeake Landfill.

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significantly affected the water levels measured at site A, and to a lesser extent influenced water heights measured at well C_1 . This fluctuation in the hydraulic head along transect A-C resulted in a regular fluctuation of the gradient along that transect. The gradient toward the river was at a minimum at high tide (0.004) and at ^a maximum at low tide (0.008). At no point in the tidal cycles did the gradient reverse itself. From these data, it appears that the rate of groundwater flow toward the river will change over ^a tidal cycle. It also appears that throughout a complete tidal cycle, net flow of groundwater remains in a northerly direction.

CHEMICAL ANALYSES

Simple statistics such as means and standard deviations, as well as one way ANOVAs, are used as an aid in interpreting the results. All ANOVA tests used F=0.01 as the rejection limit for the null hypothesis. Heterogeneity of variance between sample sites for every variable was the most serious failure of an assumption for ANOVA. This non-normal distribution of errors was ^a direct function of its mean value $(S^2=m+S^2_m)$. In an attempt to correct for this failure, ^a lognormal transformation was used for each variable. After the data was lognormally transformed, homogeniety of variance was improved, though not enough for the variance to be normally distributed (Barlett-Box ^F statistic). Using lognormally transformed data did not signifcantly change results of the ^F statistic for ANOVA

over the original data. The original data set was used in subsequent discussions because of the controversy over applying ^a lognormal transform to ^a pseudo lognormal distribution; the failure of the lognormal transform to significantly improve homogeneity of variance; and the unbiased estimate provided by non-transformed data for sample means and standard deviations. Use of nontransformed data, even though the assumptions of homogenety of error variance and normally distributed error failed, is supported (in ^a qualified way) by Cochran (1947). Cochran noted that non-normality and heterogenity of errors often do not greatly effect the validity of the F-test. He does point out, though, that the results should be regarded as approximative rather than exact.

pH

Values for pH varied significantly between sites, ranging from 4.70 at well I_1 to 7.45 in the channel. From Figure 10, an obvious pattern of increasing pH with depth for every well site can be seen. To test if this increase was significant, ^a series of Student's T-tests were used, comparing the upper wells with their corresponding lower well. For every site (excluding ^B and C) the null hypothesis that the two means were equal was rejected at the 0.01 significance level. The increase in pH as well as higher concentrations of Ca with depth (see calcium, pg. 43) was probably caused by increased shelly material with depth. The higher pH at sites ^B and ^C relative to the

Bar graph of site means for pH, vertical lines represent 95% confidence ranges. Figure 10.

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corresponding lower wells was likely due to leachate movement from the landfill. The clay lens separating upper from lower well at site ^C would prevent movement of leachate vertically.

Well B_1 appears to show some seasonal cyclicity in pH; higher in the summer months and lower in the winter. The cause of cyclicity (also observed in several other parameters at B_1) is attributed to leachate migration. Microbial decomposition of organic wastes would increase due to the rise in temperature and rainfall during the summer months. The sandy nature of the landfill cover allows for quick infiltration of rainwater, and the high water table (above the landfill's base) allows for direct contact between the leachate and groundwater. Rapid infiltration of rainwater and the high water table appears to overide the effects of increased evapotranspiration. This suggestion cannot be confirmed until water budget approximations are made for the area around the landfill.

Eh

Eh ranged from -210 mv at site G₂ to $+255$ mv at site I₁. Both sample means and means by sample period (date) varied significantly. Samples taken during the period between June and July appear to be significantly more oxidizing than the other dates (Figure 11). Winter samples are on the average more reducing, with a minimum for the December

Figure 11. Bar graphs of site means for Eh and temperature, vertical lines represent 95% confidence ranges.

sampling. Site I_1 is the most oxidized site (X=91 $+71$ mv) and F_1 the most reduced (X=-118 \pm 68mv).

Temperature

Temperature, though not considered to be a very sensitive measure for presence of leachate, was included early (10-2S-82) in the research only because it was required for determining salinity and was already available. Data missing for the period 12-20-82 was due to instrument problems, resulting in the additional loss of conductivity and salinity measurements.

Temperature for the well sites varied from 9^oC for sites B_1 and F_1 in January, and B_2 in March to 25^OC at site C_1 in July. The surface water sites, as expected, had a larger range, from 6^oC in January to 31^oC in July. There was significant variation in both site and sample period means. Temperature for the upper wells were generally higher than lower wells, though not significantly. Variation in temperature by date was seasonally cyclic, with a low of 11° C and a high of 21° C (Figure 11).

Conductivity and Salinity

There was a wide variation in conductivity between sites, ranging from 100 umohs at site I_1 and I_2 to 22,000 mohs in the river. Salinity varied from % for the control wells to 14% in the surface water. Mean conductivities and salinities are graphed on Figure 12. Site means for both conductivity and salinity were significantly different. The control wells all have mean conductivities at least an

Bar graphs of site means for conductivity and
salinity, vertical lines represent 95% confidence Figure 12. ranges.

order of magnitude less than the other sites. There is ^a significantly sharp decrease in salinity and conductivity away from the landfill for transect $A_1 - C_1$ and a slight U-shaped trend along transect D_1-F_1 , decreasing away from both landfill and river. Though this decrease along transect D_1-F_1 is not statistically significant, locally it does suggest both brackish surface water and leachate might influence groundwater salinity and conductivity. Well G_2 had a significantly higher concentration than G_1 , probably due to a density-separated flow of leachate from the landfill. The large variation in salinity and conductivity at B_1 results from seasonal influence on the leachate plume. Concentrations for the summer months are significantly higher than for the winter months.

Nitrate and Nitrite

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Both nitrate and nitrite were present in low concentrations. Nitrate varied from a maximum of 1.0 mg/l at J_1 in October to BDL $(\langle 0.1 \text{ mg}/1)$ for all sites. Nitrite had a maximum concentration of ⁸⁵ ug/1 for the channel in October and a minimum of BDL (<1 yg/1) at most sites. Nitrate's mean by site was less than 0.3 mg/1 and nitrite's was less than 10 μ g/1 (except the channel, with a mean of 32 μ g/1). Nitrate tended to be higher in concentration for sites C_1 and C_2 , I_2 , J_1 , and surface water (Figure 13). Nitrite on the other hand was much higher in the channel (with ^a correspondingly larger variance).

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Figure 13. Bar graphs of site means for nitrate and nitrite, vertical lines represent 95% confidence ranges.

Total Kjeldahal Nitrogen

Total Kjeldahal Nitrogen (TKN) varied by over three orders of magnitude between sites (Figure 14). TKN decreased significantly away from the landfill for both transects followed by G_2 then G_1 . The higher concentration in $A_1 - C_1$ and $D_1 - F_1$. Site C_1 (adjacent to the active portion landfill) had by far the highest concentration of TKN, significantly away from the landfill for both transects
A₁-C₁ and D₁-F₁. Site C₁ (adjacent to the active portion
the landfill) had by far the highest concentration of TKN,
followed by G₂ then G₁. The higher separated flow of leachate from the landfill. Concentration of TKN in all surface waters was negligable. Well C_2 had a relatively high mean (58 mg/1) due to the anomalously high concentration from the first sample date. This anomalous value resulted from leakage through the clay lense separating C_1 from C_2 , while drilling well C_2 in August 1982. Site B_1 may show seasonal variability in TKN, but unfortunately, the data set is incomplete. Data missing for dates 10-28-82, 6-3-83, and 8-28-83 was the result of problems with the ammonia probe. The analysis for 3-6-83 was not conducted within the alloted holding time.

Total and Orthophosphate

Orthophosphate accounted for a majority of the phosphate in the ground and surface waters (Figure 15), averaging greater than 50% for all sites. Total phosphate, however, was low for all surface and groundwater sites, ranging from less than 0.01 to 2.4 mg/l at C_1 . Sites most likely to be influenced by leachate $(C_1$ and G_2) had the highest averages

Figure 14. Bar graph of site means for total kjeldahal nitrogen, vertical lines represent 95% confidence ranges.

Figure 15. Bar graphs of site means for total phosphate and orthophosphate, vertical lines represent 95% confidence ranges.

of all sites, in most cases at least an order of magnitude greater than the other sites.

ANOVA indicated at least one site mean was significantly different from all others. Tukey's Studentized range test by site separated C_1 and G_2 from the other sites for both total and orthophosphate. B₁ and G₁ separated from other sites, with very little overlap, for orthophosphate.

Site B_1 has a relatively high mean and standard deviation (Appendix D), and on inspection of the total data set, there may be some seasonal variations at this site for both total and orthophosphate. Student's T-test for the summer samples (July and August) against the other dates indicated that the summer concentrations were significantly higher than winter concentrations. This increase in phosphate at site B_1 is interpreted as an increase in leachate production during the summer months.

Phosphate was found in significantly greater concentrations at sites C_1 , G_2 , B_1 and G_1 . The higher concentration for the lower well at site ^G may indicate ^a density separated flow of leachate from the landfill. Phosphate in the surface waters was low (averaging less than 0.01 mg/1).

Sulfate

Sulfate concentrations were relatively low for the groundwater samples, averaging ¹⁵ mg/1, while surface water sites were over an order of magnitude higher, averaging ⁴⁵⁴ mg/1 (Figure 16).

Bar graph of site means for sulfate, vertical
lines represent 95% confidence ranges. Figure 16.

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ANOVA rejected the null hypothesis that all site means were equal, and Tukey's range test significantly separated surface water from groundwater sites. The range test also separated the channel and river sites. Separation of the channel site from the river site was due to the anomalously large difference in measured sulfate concentrations for the sample taken on 10-30-83. Removal of this sample resulted in no significant difference between the channel and river sites. The heterogeneity within the channel and river sites may result from tidal fluctuations or use of the locks. This source of variance in the surface water samples was not supported by regression analysis.

Sites adjacent to the river (A_1, A_2, D_1) had higher concentrations of sulfate than the other groundwater sites, indicating influence from the surface water. Sites ^I and J also had elevated sulfate concentrations with the source possibly from nearby drainage ditches. The increased sulfate concentrations are not statistically significant, indicating only a trend toward higher concentrations.

Sodium and Chloride

Sodium concentration equaled chloride concentration for all sites (Figure 17; Appendix D). Site means were significantly different, with Tukey's range test separating surface from groundwater for both sodium and chloride. Groundwater sites down gradient of the landfill had much higher concentrations (at least an order of magnitude) than sites up gradient (control sites) of the landfill. Sodium

Bar graphs of site means for sodium and chloride,
vertical lines represent 95% confidence ranges. Figure 17.

and chloride concentrations between the river (and channel) and landfill appear to be influenced by both the surface water and leachate. This dual influence is best illustrated along transect A-C (river to landfill) where the chloride concentration decreases toward the landfill for lower wells and increases for upper wells. The large variability at site B_1 is probably seasonally influenced, with higher concentrations in the summer months and lower concentrations in the fall through spring months (total data, Appendix C). This variability is probably due to increased leachate movement, since the surface water does not appear to significantly increase seasonally. The inverse relationship (also seen in conductivity and salinity) between upper and lower wells along transect A-C possibly represents a density separated flow, in which a denser saline wedge from the surface water extends toward the landfill and a less dense leachate plume overides the saline wedge.

Potassium

Potassium varied considerably from site to site, with ^a minimum of 1 mg/1 at site H_1 to a maximum of 1530 mg/1 at site C_1 in August. Site means varied significantly, and Tukey's range test separated C_1 and G_2 , then G_1 and B_1 from the other sites. All other sites fell within the same range, including surface waters (Figure 18). Sites adjacent to and down gradient (A-G) of the landfill have means at least an order of magnitude greater than the other sites. Variability at B_1 is almost certainly seasonal, increasing

Bar graph of site means for potassium, vertical
lines represent 95% confidence ranges. Figure 18.

by over an order of magnitude in the summer months (July and August). An increase in June and decrease in September and October is also apparent (Appendix (C)).

The sharp decrease in potassium along transect A-C is undoubtably due to movement of leachate. Higher levels of potassium for the lower well ^G would result from ^a density separated flow of leachate.

Calcium

Calcium varied significantly between upper and lower well means (based on Student's ^T test). The upper wells mean was $54 + 36$ mg/1; the lower wells mean was $108 + 55$ mg/1, and that of the surface waters was 121 ± 45 mg/l. For every groundwater site, the lower well had ^a higher concentration of calcium (Figure 19). This increase in calcium with depth is caused, in part, by the increase in shell material with depth. The upper wells have higher concentrations of calcium downgradient of the landfill (sites A-G) than upgradient (control sites H-J). The higher concentration of calcium at these sites may be influenced by a combination of leachate and surface water.

Though Tukey's range test did not separate any groups without overlap (except site F_2), the general trendency was for surface waters and lower well sites together with higher means, while upper well sites had lower means.

Magnesium

The concentration of magnesium is significantly higher (Student's ^T test) for surface waters than groundwaters.

Bar graphs of site means for calcium and magnesium, vertical lines represent 95% confidence ranges. Figure 19.

Surface waters had a mean concentration of $335 + 137$ mg/1, while the groundwater samples had a mean of $46 + 48$ mg/l (Figure 19). Tukey's range test separated surface waters from groundwater without overlap. Within the groundwater sites, G_2 and C_1 separated from the other sites with very little overlap. Although the control wells did not separate from the other sites without overlap for Tukey's range test, they all had much lower means. The inversely trending concentrations between upper and lower wells along transect A-C for Na and Cl was also true for magnesium.

Iron and Manganese

Neither iron or manganese varied significantly between sites. However, iron did vary significantly over time. From inspection of the total data set (Appendix C), it can quickly be seen that for several sites $(F_1, A_2, D_2, F_2,$ G₂, I₂ especially), the measured concentration for the first sampling period was several orders of magnitude greater than the following dates. These anomously high concentrations were the result of an inadvertent partial acid extraction of sediment not filtered out of the samples. Suspended particulates in subsequent samples were removed by centrifuging.

Ry removing the first sample date for both iron and manganese, variance between sites became significant, whereas variance between dates was no longer significant. Although there is no strong trends for either element, the upper wells generally have higher concentrations, with the

exception of sites E_2 and I_2 . High iron and manganese concentrations at these two sites is attributed to limited flushing of the well prior to sampling necessitated by low recharge rates.

Zinc

Zinc concentrations varied significantly by site. The null hypothesis was not rejected for ANOVA by site. Because the same sites which had anomously high iron and manganese concentrations $(F_1, A_2, D_2, F_2, G_2$ and I_2) also had high zinc concentrations, the first sample date was removed and ANOVA reexecuted. Site ^A wells (both upper and lower) give the only two mean concentrations significantly different from the others. A_1 had a higher concentration due to the value for date $3-6-83$ (0.39 mg/1), while A₂ had consistantly higher concentrations than the other sites. A₂ was the only site to separate from the other sites by Tukey's range test.

FACTOR ANALYSIS

Principal components analysis was initially run on the data. Out of the twenty vectors extracted, five accounted for over 80% of ihe variance in the data, and seven accounted for over 90% of the variance (Table 2). Next, a principal axes solution was applied to the correlation matrix. Because of high communalities for several variables, the diagonal element was not replaced by communality estimates. ^A varimax procedure was used to

Table 2. Table of eigenvalues for the principal components method, before and after VARIMAX rotation.

EIGENVALUES

PC METHOD WITHOUT ROTATION PC METHOD WITH VARIMAX ROTATION Factor Eigenvalue % Variance 1 2 7.13 3.61 35.6 $\overline{18}$ Factor 1 2 Eigenvalue 4 Varience 6.56 4.02 39.3 24.1

TABLE 3. Table of factor loadings for the principal rasic of factor followings for the principal
components method, before and after VARIMAX
rotation.

Factor Loadings without Rotation

Rotated Factor Loadings

rotate the vectors, reducing the number of factors representing the data's variance from twenty to five loadings. These factors were then compared to loadings of the corresponding factors from principal components analysis. Since there was no major change in factor loadings and scores, the rotated matrix solution was used.

Factor One (Surface Water — Ground Water)

Variance accounted for by factor one was mostly between groundwater and surface water sites representing 39% of total variance in the data. Loadings, or variables, most important in determining the direction of the vector (i.e.: largest magnitude) would exhibit the most variance between the groundwater and surface water. These loadings were conductivity, salinity, hardness, chloride, sodium, and magnesium. Scores for factor one were calculated and grouped according to site (Figure 20).

The channel and river sites had significantly higher score means than the total score mean and any of the ground water sites. All ground water sites, with the exception of G₂, had means below the total mean. Because all maximum loadings were positive, it would be safe to assume that conductivity, salinity, hardness, chloride, sodium, and magnesium are present in much higher concentrations in the surface water than ground water. This observation is supported by the raw data for these parameters.

Also worth noting is the inversely related trends between upper and lower wells along transect A-C. This

Bar graphs of factor loadings and score means by Figure 20. site for Factor 1, vertical lines represent 95% confidence ranges.

general increase in concentrations away from the river for the upper wells and decrease away from the river for the lower wells was present for many of the parameters listed earlier (hardness, sodium, chloride, and magnesium). The control wells (H, I, and J) all had the lowest mean scores among all sites.

Factor Two (Leachate Influence)

Factor two delineated those parameters most indicative of leachate from the landfill. Variables with the greatest magnitude along factor two were TKN, total phosphate, orthophosphate, and potassium. These four variables account for, approximately, 24% of the variance in the data.

Those sites expected to be influenced most by leachate $(C_1$ and $G_2)$ had significantly higher mean scores (Figure 21). Score means for B_1 and G_2 were also, as expected, higher than the average.

Channel and river sites grouped with the other ground water sites, with mean site scores well below the total score mean. The large variance at B_1 is due to seasonality with summer samples (June-August) having much higher scores than the fall through spring scores.

TKN, total phosphate, orthophosphate and potassium are found in high concentrations in the leachate and in much lower concentrations in the surrounding ground and surface waters.

Figure 21. Bar graphs of factor loadings and score means by site for Factor 2, vertical lines represent
95% confidence ranges.

Factor Three (Sample Technique)

Factor three reflected a problem encountered in the first sampling period (9-2-82). ^A portion of the samples were field preserved with 1:1 HNO₃ for metal analysis. Several sites had ^a large amount of suspended particulates (especially F_2 and I_2), and consequently, the particulates were partially extracted by the acid. This resulted in anomalously high metal concentrations (particularly iron, manganese, and zinc) for the first sample period. For subsequent sampling, suspended material was removed by centrifugation prior to addition of the acid.

Since these anomalously high metal concentrations constituted a known source of error in the data, it was to investigate how much variability was added ted a known source or error in the data, it was o
to investigate how much variability was added by
n of the first sampling period. This would suppl
ative estimate of the actual importance of other inclusion of the first sampling period. This would supply a qualitative estimate of the actual importance of other sources of variance. This source of error was later removed and the data re-analyzed with the principal factor technique.

The variables with high loadings for factor three were iron, manganese, zinc and to a lesser extent calcium and hardness (Figure 22). Sites F_2 and I_2 had, expectedly, very large variances due to the suspended sediment in the sample for the first date. Sites which typically had the least amount of sediment had the smallest variances.

Because this source of variance was due to sampling technique, analyses for the first date was removed from the data and factor analysis rerun. The result was to shift the

Figure 22. Bar graphs of factor loadings and score means by site for Factor 3, vertical lines represent 95% confidence ranges.
variance explained by factors four and five to factors three and four, with little change in factors one and two. With the removal of factor three, which accounted for 19% of the total variance, factors one and two increase in proportion of variance from 39% to 44% and 24% to 27% respectively. Factors four and five increased by less than 1% of total variance explained. The large increase in variance accounted for by factors ¹ and ² (13% for each factor) emphasizes the importance of these two factors over the remaining factor.

Factor Four (Site Depth)

Factor four accounted for 10% of the total variance and separated upper (3 meter) from lower (10 meter) wells. Eh, pH, temperature, and calcium had the largest magnitudes along this vector (Figure 23). Calcium and pH were negative loadings, indicat'ing an inversely related trend between the original data and corresponding factor scores. In all cases, on ^a site by site basis, the upper well mean was greater than the lower well. However, this can only be considered a trend, as the difference between upper and lower wells was not significant for all sites. This trend suggests a general increase in pH and calcium and decrease in temperature and Eh (more reducing) with depth.

Factor Five (Remaining Variance)

Factor five accounted for the remainder of the variance the data. Nitrate and nitrite were the principal sources

Figure 23. Bar graphs of factor loadings and score means by site for Factor 4, vertical lines represent 95% confidence ranges.

of variance, reflecting the lower means at sites D_2 , E_2 , F_2 and G_2 . This factor also accounted for large amount of variance within surface water sites as well as sites A_2 , B, C and J_1 for nitrate and nitrite (Figure 24).

The variance accounted for by factor five is relatively minor (8% of the total variance) and is not as well reflected by the original data as the other four factors. Therefore, only factors one through four (excluding three for sample handling error) accounted for geochemically interpretable variance. Factor five accounts for the remaining, relatively minor, variance in the data.

Factor Analysis Summary

Sixty-three percent of variance in the data was attributed to influence from surface water (39%) and leachate (24%). Conductivity, salinity, hardness, chloride, sodium, and magnesium had significantly higher concentrations in the surface water than ground water. These parameters were also present in elevated concentrations in the leachate. Leachate,

however, was best characterized by high concentrations total phosphate, orthophosphate, and ever, was best characterized by high concentrations of
total phosphate, orthophosphate, and potassium. TKN,
sphates, and potassium were present in much lower conce
ions in the surface water and ground water. Calcium
pH inc phosphates, and potassium were present in much lower concentrations in the surface water and ground water. Calcium and pH increased with well depth due to an increase in shell material.

Figure 24. Bar graphs of factor loadings and score means bar graphs of factor foadings and score means
by site for Factor 5, vertical lines represent 95% confidence ranges.

REGRESSION

Regression analysis was used to investigate the influence of tidal fluctuation, seasonal variation, depth, and horizontal distance on the parameters measured in this study. The model including all independent variables accounted for greater than 50% of the total variance for only a few dependent variables (Table 4). These were (in order of importance) chloride, sodium, magnesium, conductivity, temperature, hardness, calcium and salinity. Only chloride accounted for greater than 75% of the total variance.

Tidal fluctuation, independent of the other variables, accounted for less than 10% of the total variance for all dependent variables. Depth as the independent variable accounted for greater than 50% of total variance for chloride, sodium, and magnesium. Temperature was the only dependent variable which seasonal variation accounted for greater than 50% of the variance. Depth was the best estimator (regressor) for chloride, sodium, magnesium, conductivity, hardness and salinity; accounting for 70%, 62%, 54%, 45%, and 40% of the variance, respectively. Seasonal variation was the best estimator for temperature (55%) and horizontal distance for calcium (33%).

The variables best estimated by the independent (regressor) variables closely matched the variables with high loadings along factor one (surface water/groundwater). Only temperature and calcium did not have high loadings for

TABLE 4. Coefficient of determination (r^2) for five
multiple regression models (depth, distance,
seasonal, tidal, and total)

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factor one. All the variables with high factor loadings (conductivity, salinity, hardness, sulfate, chloride, sodium, and magnesium) were most dependent on depth as a regressor variable(s). Since the three depths possible were surface water, ³ meter wells, and ¹⁰ meter wells, this dependency was not surprising.

As a check for this observed relationship between the dummy variables for depth and high loadings along factor one, all independent variables together, as well as separately, were regressed against the factor scores for factors one and two. The independent variables together accounted for 73% of the variance in factor one, with depth responsible for 61% of the variance. For factor two (leachate/groundwater-surface water), 37% of the variance was accounted for by the regressor variables. Horizontal distance was responsible for 33% of the variance in factor two scores. Variables with high loadings along factor two (TKN, TP04, OP04 and K) were influenced most by horizontal distance. With the greatest decrease in leachate concentration occuring horizontally between the landfill and river $(transect A-C)$, the relationship between factor two scores and horizontal distance was expected.

The independent variable of particular interest in this regression analysis (tidal fluctuation) was neither statistically or geochemically significant. To further investigate possible tidal influence on the groundwater geochemistry, conductivities and water levels were measured

hourly along transect A-C for one complete tidal cycle. There was ^a slight correlation between conductivity and water level for the upper wells, and no correlation for the lower wells. At site ^A (adjacent to the river), the upper well (A_1) had a correlation of 0.63 and the lower well (A_2) had ^a correlation of 0.05 (Figure 25). Conductivity correlated with water level at 0.48 for well C_1 and at ≤ 0.01 for well C_2 . The slight correlation for the upper wells (A_1) and C_1) suggests there may be some minimal influence from tidal fluctuation. When conductivity was correlated with tidal level over the entire year at A_1 , there was no significant inter-relationship (r=0.03). Although tidal fluctuation has no significant long term impact on the parameters measured, correlation between conductivity and water level at well A_1 is additional evidence that may be some intrusion of brackish water from the river.

SUMMARY AND RECOMMENDATIONS

The Chesapeake landfill appears to be typical of many municipal landfills in coastal plain and other low-lying areas. The landfill is located in, and bounded to the north and west by a tidal marsh. The Chesapeake landfill first began operation in the mid to late sixties, and is therefore ^a relatively old (mature) landfill. The original mode of operation was to trench and dewater while the trash was deposited and compacted. Presently, refuse is being redepositted over the oldest portion of the landfill

Figure 25. Graphs of tidal fluctuation and conductivity with time for wells Al and A2.

(eastern edge), extending as far west as transect A-C (Figure 5, p. 12).

The water table is high in this area, as evidenced by the adjacent marsh, and the landfill's base is in ^a medium to fine sand which underlies the marsh clay. Since no liner, clay or otherwise, was initially used, it would be safe to assume the landfill is at least partially saturated with groundwater. Cover for the landfill is of local material and usually sandy in nature. The refuse is primarily from residential and small industrial sources, with the largest industrial input being wood and wood by-products. With no real physical impediment for leachate flow from the landfill, the low hydraulic gradient would be the primary restrictor of leachate movement. The leachate plume is concentrated north of the landfill, centered arount transect A-C and paralleling the hydrologic head distribution (Figure 8). Leachate found to the west, at site G, is due to the close proximity of the surface water (which abuts the landfill at high tide). Net flow of groundwater around site G is probably west. Any detrimental effect from leachate on the surrounding ground and surface waters would be from nutrient enrichment of the adjacent marsh and waterway. It has been suggested, however, that the marsh may serve as both a nutrient source and sink, especially for nitrogen and phosphorus (Heinel and Flemer, 1976; Valeela et al., 1978; Wolaver et al., 1983; Wolaver and Zieman, 1984). If this is true, the leachate may have

little detrimental influence on the adjacent surface water. Interaction of leachate with the marshland deserves further study.

The source of variance in many parameters measured in this study was traced back to either surface water or leachate influence. Higher concentrations for hardness, sodium, chloride, magnesium, and sulfate, as well as, high conductivity and salinity was indicative of the surface water. These seven variables (except sulfate) were also present in high concentrations in the leachate, relative to the ambient groundwater. Well sites adjacent to the surface water had elevated concentrations for hardness, sulfate, sodium, chloride, and magnesium. This is probably due to some recharge from the river.

Total kjeldahl nitrogen, phosphate (total and ortho), and leachate than ambient groundwater and surface water. Concentrations for these parameters did not differ significantly between the ground and surface water.

Organic nitrogen and phosphates are very common byproducts of biochemical-chemical degredation of municipal refuse. They possess qualities desirable for a good leachate indicator (high concentration in the leachate and mobile). However, because of interferences with the analysis of phosphates and nitrogen, a great deal of care must be taken in the labratory procedures. Total kjeldhal nitrogen was particularly difficult to measure.

Potassium, on the other hand, is also present in leachate in high concentrations and is relatively mobile with few, and minor, interferences. The ease in preservation and long holding time makes potassium additionally attractive as a leachate indicator. It also is a common by-product of organic waste degredation and has been recognized as a primary constituent of leachate, especially during generation and stabilization of the fill (Chain and De Walle, 1975; Johansen and Carlson, 1976; Fenn et al., 1977; Ellis, 1979; Lu et al., 1981; Tredoux, 1984).

Temperature, Eh, pH, and calcium varied most with depth. Temperature decreased and Eh became more reducing while calcium and pH increased with depth. The decrease in temperature and Eh with depth follows the expected natural trend for a water table aquifer. The increase in calcium and PH was due, at least in part, to an increase in calcareous shell material with depth.

While tidal fluctuation had no observable influence on the groundwater geochemistry, there did appear to be a significant seasonal influence for most parameters measured. Salinity, conductivity, pH, total kjeldhal nitrogen, total and orthophosphate, chloride, sodium, potassium, calcium, and magnesium increased sharply during the summer months for site B_1 (midway between the landfill and river). Ideally the wells should be monitored monthly for the summers (June through August) and seasonally for the remaining year (fall, winter, and spring) in order to

establish the exact affect of seasonal change on leachate variability. ^A rain gauge should be installed to monitor local precipitation, and evapotranspiration estimated for water budget approximations. In order to accurately predict seasonal influence, monitoring would have to take place over several complete seasonal cycles. Parameters recommended for measurement in such a study are pH, conductivity, orthophosphate, sodium (or chloride), potassium, and calcium.

CONCLUSIONS

Variation in most parameters measured in this study was due to either leachate influence or surface water influence. Leachate was characterized best by significantly higher concentrations of potassium, total phosphate, orthophosphate, and total kjeldahl nitrogen over the ambient groundwater and surface water. Conductivity, salinity, hardness, sodium, chloride, and magnesium were also present in leachate at significantly higher levels than the ambient groundwater. Although higher pH and calcium values were also indicatative of leachate, the major source of variance for these parameters was apparently due to an increase in calcareous shell material with depth.

Parameters found in significantly higher levels in the surface water over leachate and groundwater were conductivity, salinity, hardness, sodium, chloride, magnesium, and sulfate. groundwater samples taken adjacent to the surface water had elevated concentrations for hardness, sulfate, sodium, chloride, and magnesium. This is most likely due to intrusion of brackish water from the river.

Because of influence from sources other than landfill leachate, parameters traditionally used for routinely monitoring leachate (pH, chloride, and conductivity) are not appropriate for sites in, or adjacent to naturally

brackish water. Potassium appears to be the most promising parameter for routine monitoring in coastal marshlands and estuaries, in that it is:

- 1) present in leachate in much higher concentrations than the ambient groundwater
- 2) easily stored, preserved, and analyzed
- 3) relatively free of interferences

Total phosphate and orthophosphate are also good candidates for use in routine monitoring, although greater care must be taken to correct for interferences for these analyses. TKN, though present in much higher levels in the leachate, is too time consuming to measure to be used routinely.

Tidal fluctuation had no apparent influence on the groundwater geochemistry. Seasonal variation, however, did affect salinity, conductivity, PH, total kjeldhal nitrogen, potassium, calcium, and magnesium at site B_1 (midway between the landfill and river). These parameters increased significantly during the summer months (July through August). Additional monitoring is required to accurately evaluate the effect of seasonal variation on the leachate. Lateral variation in groundwater mounding under the landfill and heterogeneity in grain size distribution in the aquifer significantly influenced the position of the leachate plume over a short areal distance . This emphasizes the need for a rigourous hydrologic evaluation of the aquifer in contact with the landfill before installation of wells for routine monitoring.

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Appendix A. Well logs taken from wash borings and measured well logs taken from wasn boring
water levels for all well sites.

SITE ^A

 $\mathbf{v} = \mathbf{v}^{\top}$, $\mathbf{v} = \mathbf{v}$

SITE F

CONDUCTIVITIES AND WATER LEVELS MEASURED HOURLY TO SEMIHOURLY FOR WELL SITESD ^A AND C, OCTOBER 6-7, 1982

WELL SITE ^A

Conductivity (uMOH)

WELL SITE C

Conductivity (uMOH)

Time Well 1 Well 2 1720 1820 1920 2020 2120 2220 2320 2420 0120 0220 0320 0420 0520 0620 0720 1050 2510
1000 2400 1000 2400
1030 2500 1030
1100 2490
2490 1030
1110 2500
2500 1100 2500
1150 2490 1150
1120 2470
2490 1120
1100 1100 2450
1100 2430 1100 2430
1050 2480 2480
2460 1100 2460
1100 2500 1100

Time Well ¹ Well 2 0815 3800
0910 3890 0910 3890
1015 3900 1015
1115 3900
3890 1205 3890
1310 3900 1310 1410 3880
1510 4650 1510 4650
1605 4220 1605
1710 1710 4130
1810 3950 1810 3950
1915 3980 3980
4000 2010 4000
2115 4150 2115
 2215 3910 5100 5200 5300 5500 5200 5100 5100 5100 5100 5100 5100 5100 5100 5100 5100

Water Levels (in) Water Levels (in)

Time 1706 1730 1812 1830 1910 1930 2005 2030 2109 2129 2207 2226 2301 2325 2406 2428 0102 0130 0201 0231 0301 0330 0400 Well ¹ 4 [~] 55 4. 44 4. 28 4.25 4.18 4.20 4.22 4.23 4.32 4.38 4.49 4.55 4.71 4.82 4.97 5.03 5.11 5. 11 5.10 5.06 4.94 4.85 4.74 Time Well 2 1703 3.87
1732 4.00 1732 4.00
1713 3.27 1713 3.27
1831 3.14 1831 3.14
1911 2.65 1911 2.65
1931 2.95 1931 2.95
2006 3.05 2006 3.05
2032 3.12 2032 3.12
2112 3.33 ²¹¹² ³ '3 2130 3.46
2209 3.75 2209 3.75
2227 3.90 2227 3.90
2303 4.29 2303 4.29
2327 4.50 4.50 2408 4.80
2430 4.91 4.91
 5.07 0104
 0131 0131 5.02
0202 5.01 5.01
 5.14 0233 0303 4.66
0331 4.39 4.39
 4.14 0401 Time Well ¹ 0806 6.35
0835 6.33 0835
0908 0908 6.33
0935 6.30 6.30
 6.32 1012
1035 6.31
 6.32 1108 6.32
1132 6.35 1132
1201 1201 6.32
1233 6.34 1233 6.34
1305 6.36 1305 6.36
1336 6.41 6.41
 6.42 1406
1433 1433 6.41
1505 6.44 1505 6.44
1535 6.45 1535 6.45
1603 6.45 6.45
 6.47 1634
1708 1708 6.43
1736 6.44 6.44
 6.42 1805
1835 6.43 1911 6.43 Time 0807 0838 0907 0935 1015 1037 1109 1133 1202 1235 1306 1337 1407 1434 1506 1537 1604 1635 1709 1737 1806 1836 1912

Well 2

3.03 3.06 3.18 3.34 3.58 3.82 4.13 4.36 4.62 4.86 5.07 5.19 5.24 5.20 5.10 4.93 4.77 4.53 4.27 4.06 3.79 3.54 3.34

Water Levels (in) (Cont'd)

 $We112$

3.10 3.05 3.05 3.09 3.20 3.29 3.41

WATER LEVELS MEASURED FOR ALL WELL SITES

OCTOBER 23, 1982

Appendix B. Data from previous research, Chesapeake Landfill.

CHESAPEAKE LANDFILL DATA (Rule, 1979)

NON-METAL PARAMETERS (August, 1978)

CHESAPEAKE LANDFILL DATA (Rule, 1979)

METAL CONCENTRATIONS (mg/1) (July, 1978)

METAL CONCENTRATIONS (mg/1) (August, 1978)

 $\ddot{}$

 $***NR$ No Reading

**NR No Reading

SITE 9/20/82 9/30/83 9/30/83 12/83/83 9/4/83 1/29/83 3/6/83 3/6/83 3/6/83 3/6/83 10/30/83 10/30/83 10/30/83 10
Site 8/63/83 10/30/83 10/30/83 10/30/83 10/30/83 10/30/83 10/30/83 10/4/83 10/4/83 10/4/83 10/4/83 10/4/83 10/

 $*ND = No Data$

82

Appendix C. Total data set
samples sites, for all parameters and
Chesapeake Landfill.

 $*ND = No Data$

 $\sim 10^{11}$ km s $^{-1}$

 $\mathcal{L}(\mathcal{$

 $*ND = No Data$

 \sim \sim

CONDUCTIVITY (mohs)

 $*ND = No Data$

 $*ND = No Data$
$N0_3$ (mg/1)

 $*ND = No Data$

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 $*ND = No Data$

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 TKN (mg/1)

 $\mathcal{L}(\mathcal{L})$ and $\mathcal{L}(\mathcal{L})$. The set of $\mathcal{L}(\mathcal{L})$

 $*ND = No Data$

 $\mathcal{L}^{\text{max}}_{\text{max}}$, where $\mathcal{L}^{\text{max}}_{\text{max}}$

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$0P0₄$

$S0_4$ (mg/1)

$CL (mg/1)$

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SITE				9/2/82 9/30/82 10/28/82 12/20/82 1/29/83 3/6/83 4/17/83 6/3/83 7/1/83 7/29/83 8/28/83 10/30/83								
					63	61	64	70	61	63	62	53
A ₁	660	42	38	42								
B_1	38	130	110	66	41	33	33	120	1500	1520	1500	110
c ₁	1470	1210	1260	1330	1250	1470	1390	1030	1480	1510	1530	1090
\mathbf{p}_1	16	8	7	35	28	42	45	33	33	42	39	37
E_1	72	35	29	35	57	89	94	140	200	160	140	130
F_1	74	54	48	48	60	44	54	69	81	92	87	87
G ₁	540	420	420	420	460	43	450	430	520	500	4440	420
H_1	$\mathbf 2$	\bf{z}	$\mathbf 2$	1	$\mathbf{2}$	$*_{ND}$			$\overline{2}$	$\mathbf{2}$	2	2
I_1	$\boldsymbol{2}$	$\mathbf 2$	$\overline{\mathbf{c}}$		$\mathbf 2$	$\mathbf 2$	$\mathbf{2}$	2	$\frac{2}{2}$	$\frac{2}{2}$	$\mathbf 2$	$\frac{2}{3}$
J_1	$\overline{\mathbf{3}}$	$\overline{2}$	$\overline{2}$	$\frac{2}{2}$	$\overline{2}$	$\mathbf{2}$	$\overline{2}$	$\overline{2}$			$\overline{2}$	
A ₂	27	15	29	44	$*ND$	80	80	50	27	19	18	37
B ₂	16	15	14	14	130	140	140	150	18	19	18	16
c ₂	640	730	300	120	93	75	53	43	45	53	62	57
D_2	17	11	11	13	18	18	18	18	15	15	15	15
E_2	40	*ND	29	29	32	30	28	26	28	29	29	31
F ₂	76	18	3	3	18	26	13	10	13	11	13	9
G ₂	540	970	920	780	960	840	940	920	1040	1020	1090	1120
H ₂	3	3	3	3	6	5	5	5	5	5	5	6
I ₂	27	$*ND$	3	$\mathbf 2$	$\boldsymbol{2}$	$\boldsymbol{2}$	$\boldsymbol{2}$		3	$\mathbf 2$	2	3
J_2	5	$\mathbf{2}$	$\overline{2}$	$\overline{2}$	$\overline{2}$	\overline{c}	$\overline{2}$	$\overline{2}$	$\mathbf{2}$	$\overline{2}$	$\overline{2}$	3
Ch	$*ND$	120	91	74	158	65	45	106	130	180	260	170
R	$*ND$	*ND	*ND	81	160	75	55	82	120	190	170	170

 K (mg/1)

 $***ND** = **No** Data$

$\frac{\text{Ca}}{\text{m}}$ (mg/1)

 $Fe (mg/1)$

			SITE 9/2/82 9/30/82 10/28/82 12/20/82 1/29/83 3/6/83 4/17/83 6/3/83 7/1/83 7/29/83 8/28/83 10/30/83									
A ₁	3.9	3.9	3.9	7.0	9.1	7.9	11.0	3.3	6.7	7.0	5.8	5.5
B ₁	4.0	5.2	7.8	12.0	19.0	11.0	9.8	7.9	2.3	1.4	1.1	9.9
c ₁	1.8	2.6	3.7	3.0	4.5	1.5	2.6	4.1	2.3	0.3	0.3	6.5
D_1	18.0	2.6	2.4	31.0	12.0	17.0	26.0	17.0	17.0	14.0	20.0	17.0
E_1	3.3	1.7	1.3	4.4	15.0	11.0	3.3	19.0	22.0	8.1	2.3	3.6
F_1	86.0	1.6	3.0	4.0	3.6	2.7	3.4	5,4	3.6	3.6	5.3	9.6
G_1	17.0	3.0	3.1	1.2	7.6	2.6	10.0	12.0	13.0	12.0	11.0	2.5
H_1	9.0	3.1	1.8	3.5	2.4	$*ND$	2.1	2.0	2.6	2.6	2.5	2.3
I ₁	11.0	0.9	0.8	0.2	0.3	0.4	0.4	0.3	0.9	1.5	1.6	1.0
J_1	15.0	1.3	1.6	1.6	2.7	2.7	2.9	1.8	1.2	0.4	0.2	0.1
A ₂	140.0	1.4	3.4	1.3	$*ND$	1.3	1.8	1.7	1.9			
B ₂	30.0	1.2	0.8	0.6	0.4	0.3				1.7	0.4	0.4
c ₂	7.4	6.2	2.5	0.5			0.3	0.7	0.4	1.7	0.2	0.2
	120.0	0.3	0.1		0.6	0.4	0.5	0.6	0.5	0.6	0.8	0.3
D_2	9.0	$*ND$		0.7	0.7	0.6	0.2	0.1	$\langle 0.1$	1.9	0.2	0.7
E ₂			16.0	16.0	20.0	17.0	9.0	0.2	9.0	7.3	15.0	9.6
F ₂	630.0	0.6	1.7	2.0	0.2	0.4	0.6	0.3	0.1	1.2	0.4	0.2
G ₂	260.0	4.7	0.9	0.9	2.1	1.7	1.5	1.7	1.0	2.4	0.8	1.4
H ₂	13.0	0.4	0.2	0.3	0.1	0.1	0.1	0.3	0.2	0.2	0.2	0.2
I ₂	460.0	$*_{ND}$	8.2	1.2	0.3	1.0	0.1	0.1	4.3	14.0	12.0	15.0
J_2	8.2	0.1	0.1	0.1	0.4	0.1	0.1	0.1	0.2	0.1	0.6	0.2
Ch	$*_{ND}$	0.3	0.4	0.4	0.3	0.8	0.7	1.0	0.7	0.2	0.2	0.3
\mathbf{R}	$*ND$	$*ND$	$*ND$	0.4	0.4	0.6	0.7	0.9	0.5	0.2	0.2	0.2

 $\mathcal{L}^{\text{max}}_{\text{max}}$

Mn (mg/1)

 $\frac{\text{Zn}}{\text{Zn}}$ (mg/1)

												SITE 9/2/82 9/3/82 10/28/82 12/20/82 1/29/83 3/6/83 4/17/83 6/3/83 7/1/83 7/29/83 8/28/83 10/30/83
A ₁	0.01	0.01	(0.11)	0.08	0.07	0.39	0.07	0.05	0.03	0.01	0.01	0.03
B ₁	0.02	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01
c_1	0.01	0.01	0.01	0.01	0.01	0.01	0.02	0.01	0.01	0.01	0.01	0.01
D_1	0.04	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01
E_1	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01
F ₁	0.21	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01
G ₁	0.01	0.01	$\langle 0.01$	0.01	0.02	0.05	0.05	0.01	0.01	0.01	0.01	0.01
H_1	0.01	0.01	0.01	0.01	0.03	$*ND$	0.01	0.01	0.02	0.10	0.01	0.01
I ₁	0.03	0.01	$\langle 0.01$	0.01	0.01	0.01	0.01	0.02	0.01	0.01	0.01	0.01
J ₁	0.03	0.01	0.01	0.01	0.02	0.01	0.01	0.01	0.01	0.01	0.01	0.01
A ₂	0.44	$\langle 0.01$	0.30	0.35	$*ND$	$*_{ND}$	0.38	0.23	0.10	0.02	0.01	0.13
B ₂	0.08	0.01	0.01	0.01	0.01	0.02	0.01	0.01	0.01	0.01	0.01	0.01
c ₂	0.01	0.06	0.01	0.01	0.03	0.05	0.01	0.03	0.01	0.01	0.04	0.02
D_2	0.18	0.01	0.01	0.01	0.01	0.01	0.02	0.01	0.01	0.01	0.06	0.06
E_2	0.01	*ND	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01
F ₂	2,90	0.01	0.01	0.01	0.03	0.06	0.05	(0.01)	0.01	0.01	0.02	0.03
G ₂	1.30	0.02	0.01	0.01	0.02	0.02	0.02	0.02	0.01	0.02	0.01	0.02
H ₂	0.05	0.01	0.01	0.01	0.05	0.04	0.02	0.04	0.02	0.02	0.01	0.01
I ₂	1.90	$*ND$	0.03	0.01	0.02	0.02	0.01	0.02	0.01	0.01	0.03	0.02
J_{2}	0.14	0.01	0.01	0.01	0.01	0.05	0.01	0.05	0.01	0.01	0.01	0.01
Ch	$*_{ND}$	0.02	0.03	0.02	0.04	0.02	0.01	0.01	0.01	0.01	0.01	0.01
\mathbf{R}	$*ND$	$*ND$	$*ND$	0.02	0.06	0.03	0.02	0.02	0.02	0.02	0.01	0.02

Appendix D. General descriptive statistics for all
parameters by sample site, Chesapeake
Landfill.

pH

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 $\bar{\mathcal{A}}$

 $\ddot{\phi}$

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Temperature (C)

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 \mathcal{L}^{max} .

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 \overline{a}

Salinity (0/00)

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Hardness (mg/l)

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 \bullet

Nitrate $(\text{mg}/1)$

 $\sim 10^{-10}$

 $\sim 10^{11}$

Nitrite $(ug/1)$

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TKN $(\text{mg}/1)$

		i Num. Sm pl	Mean	std. Dev.	Min	Hax	Range
 Site l Type	Site						
Upper Wells	A1	81	$1 - 375$	1.0261	$0 - 000$	$3 - 2001$	3.200
	B1	71	$28 - 9141$	57.9061	2.9001	160.0001	157.100
	C1	61	425.667	249.7851	190.0001	790.0001	600.000
	D1	71	$3 - 771$	3.3331	$0 - 1001$	$9 - 5001$	$9 - 4001$
	E1	71	10.5571	12.516	1.7001	34.0001	32.300
	71	81	24.0751	16.135	5.0001	48.0001	43.000
	G 1	5١	132.4001	66.8601	49.0001	190.0001	141.000
	81	71	0.6291	$0 - 6211$	$0 - 0001$	$1 - 9001$	1.900
	11	71	0.371	$0 - 687$	$0 - 0001$	$1 - 9001$	1.9001
	J1	71	$0 - 143$	0.294	$0 - 0001$	0.8001	$0 - 800$
Lover Tells	A2	81	$0 - 3871$	$0 - 2361$	$0 - 0001$	0. 70 0 1	0.700
	B2	81	0.8621	$0 - 9381$	0.0001	2.8001	2.800
	C2	71	57.5431	116.3291	0.7001	320.0001	319.300
	D2	71	1.429	1.6941	$0 - 0001$	$5 - 100$	$5 - 100$
	E2	61	$3 - 583$	2.6761	1.400j	8.8001	$7 - 400$
	F2	71	$6 - 486$	3.113	$1 - 7001$	$9 - 7001$	8.0001
	G2	51	214.8001	121.1331	84.0001	410-0001	326.0001
	E2	71	$0 - 800$	$0 - 9831$	$0 - 0001$	$2 - 900$	$2 - 9001$
	I2	71	$0 - 4001$	0.5231	0.0001	1.5001	1.500
	J2	71	0.4001	$0 - 3791$	$0 - 0001$	$0 - 9001$	0.9001
Surface	Channel	71	1.4001	1.955	0.0001	$5 - 6001$	$5 - 6001$
I N AL EL	<i>IRiver</i> ł	61	$0 - 450$	$0 - 3331$	$0 - 0001$	$0 - 8001$	$0 - 800$
Site Type							
Upper Wells		691	54.0621	$140 - 2371$	$0 - 0001$		790.0001 790.0001
--- ا Lower Wells		691	$22 - 525$	72.5931	$0 - 0001$	410.000	410-0001
Surface Tater		131	0.9621	1.4841	0.0001	$5 - 6001$	5.6001
Total		15 I J		35.217 107.890		0.000 790.000 790.000	

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		Num. Smpl	Мезд	Std. Dev.	Min	dax	Range
ISite IT ype	I Site						
0 pp er Wells	1A1	101	$0 - 065$	$0 - 0361$	$0 - 0101$	$0 - 1301$	0.120
	B1	101	$0 - 467$	$0 - 397$	0.1701	$1 - 3001$	$1 - 130$
	l C1	101	1,6831	$0 - 352$	1.0701	2.4001	1.330
	ו ס ו	101	0.0561	0.0651	$0 - 0001$	$0 - 2201$	$0 - 220$
	E1	101	0.171	0.3801	$0 - 020$	1.250	$1 - 230$
	F1	101	0.078	$0 - 0411$	$0 - 0001$	$0 - 150$	0.1501
	G1	101	0.201	$0 - 185$	$0 - 0601$	$0 - 6701$	0.6101
	H1	101	0.0231	$0 - 0221$	$0 - 0001$	0.0701	0.070
	11	1 O J	0.012	0.0191	$0 - 0001$	$0 - 0601$	$0 - 060$
	J1	91	$0 - 032$	$0 - 0471$	0.0001	0. 15 0	0.1501
Lower Wells	A2	91	0.0601	$0 - 1431$	$0 - 0001$	0.4401	$0 - 4 + 0$
	B ₂	101	$0 - 136$	0.3741	$0 - 0001$	1–2001	1.200
	C ₂	101	0.1771	$0 - 2011$	$0 - 0301$	$0 - 6101$	0.580j
	D2	101	0.137	$0 - 3671$	$0 - 000$	1.1801	$1 - 180$ $\frac{1}{2}$
	E ₂	91	$0 - 2721$	$0 - 7231$	$0 - 0001$	$2 - 2001$	2.2001
	F2	101	$0 - 0401$	0.064	$0 - 0001$	$0 - 2201$	0.2201
	G ₂	101	$1 - 359$ J	$0 - 1861$	$1 - 100$	1.600j	0.500
	н2	101	$0 - 0251$	$0 - 045$	0.0001	$0 - 1501$	0.1501
	I ₂	91	$0 - 167$	0.4631	$0 - 0001$	1.4001	1.4001
	J2	101	$0 - 1621$	0.3301	$0 - 0301$	1.1001	1.0701
	Surface (Channel	9 ₁	0.0481	$0 - 0231$	0.0301	$0 - 0901$	0.0601
rat er	River	71	0.0401	0.0131	0.0301	$0 - 060j$	0.0301
Site Type							
 Upper Wells		. 99	0.2811	0.5321	$0 - 0001$	$2 - 400j$	$2 - 4001$
over Wells		971	$0 - 256$	$0 - 503$	$0 - 000j$	$2 - 200j$	$2 - 2001$
Surface Water		16	0.044	$0 - 0191$	$0 - 0301$	$0 - 0901$	$0 - 0601$
Total		2121	0.252	$0 - 5001$	$0 - 0001$	2.4001	$2 - 4001$

Total Phosphate (mg/1)

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 $Sulfate (mg/l)$

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Chloride $(\text{mg}/1)$

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Potassium $(\text{mg}/1)$

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Calcium $(\frac{\pi q}{1})$

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Magnesium $(\mathbb{R} q/\mathbb{1})$

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Iron $(\text{mg}/1)$

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Manganese $(\text{mg}/1)$

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